Magnetic Structural Studies of the Two Polymorphs of Li₃Fe₂(PO₄)₃: Analysis of the Magnetic Ground State from Super-Super Exchange Interactions

Gwenaëlle Rousse,*^{,†} Juan Rodríguez-Carvajal,[‡] Calin Wurm,[§] and Christian Masquelier[§]

Institut Laue Langevin, BP 156, F-38042 Grenoble Cedex 9, France, Laboratoire Léon Brillouin (CEA-CNRŠ), CEA/Saclay, 91191 Gif sur Yvette Cedex, France, and Laboratoire de Réactivité et Chimie des Solides, Université Picardie Jules Verne, 33 Rue St. Leu, 80039 Amiens Cedex 9, France

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The nuclear and magnetic structures of the monoclinic (P21/n, A-LFP) and rhombohedral $(R\overline{3} \text{ B-LFP})$ forms of Li₃Fe₂(PO₄)₃ have been solved by using powder neutron diffraction at room temperature and 1.5 K on polycrystalline samples. Both structures are built on [Fe2- $(PO_4)_3$ 'lantern units' that are connected in a different way for each form. Measurements by a superconducting quantum interference device reveal a global antiferromagnetic behavior with ordering temperatures of 25 and 23 K for the A and B forms, respectively. Both magnetic structures, determined from symmetry analysis and Rietveld refinements of neutron diffraction data recorded at 1.5 K are collinear. The magnetic moments are perpendicular to [001] in both structures. The obtained magnetic moments are 4.7 and 3.9 $\mu_{\rm B}$ per iron atom for the A form (ferrimagnetic ordering of the two iron sublattices) and 4.7 $\mu_{\rm B}$ for the NASICON (B-LFP) form. The Fe atoms are oriented antiparallel within the $[Fe_2(PO_4)_3]$ lantern units, while parallel orientation takes place between Fe atoms that do not belong to the same [Fe₂(PO₄)₃] lantern unit. Using numerical calculations we have established a magnetic phase diagram and determined the necessary constraints to be satisfied by the values of the exchange interactions to obtain the observed magnetic structures as the ground state.

Introduction

Three-dimensional polyanionic structures built of interconnected MO₆ octahedra and XO₄ tetrahedra, which provide an interstitial space partially occupied by alkali cations, have been the subject of intense research for their unusual ionic conductivity, thermal expansion, and electrochemical properties during the past 20 years. The most widely investigated is the NASICON family A_xMM'(XO₄)₃,¹ which offers extensive versatility toward chemical substitutions in the [MM'-(XO₄)₃]_∞ polyanionic framework into which between 0 and 5 alkali cations may be accommodated. Recent studies have been devoted to their interesting electrochemical properties as positive electrode materials in rechargeable lithium batteries.²⁻⁹ In particular, owing

* To whom correspondence should be addressed. E-mail: rousse@ill.fr. [†] Institut Laue Langevin.

- (1) Hong, H. Y. P. Mater. Res. Bull. 1976, 11, 173.
- (2) Okada, S.; Nanjundaswamy, K. S.; Manthiram, A.; Goodenough, J. B. Proceedings of the 36th Power Sources Conference, Cherry Hill,

- (3) Padhi, A. K.; Nanjundaswamy, K. S.; Goodenough, J. B. J. Electrochem. Soc. 1997, 144, 1188.
 (4) Nanjundaswamy, K. S.; Padhi, A. K.; Goodenough, J. B.; Okada, S.; Ohtsuka, H.; Arai, H.; Yamaki, J. Solid State Ionics 1996, 92, 1.
- (5) Goodenough, J. B.; Padhi, A. K.; Masquelier, C.; Nanjun-daswamy, K. S. U.S. Patent 08/840,523, 1997.
 (6) Padhi, A. K.; Nanjundaswamy, K. S.; Masquelier, C.; Okada, S.; Goodenough, J. B. *J. Electrochem. Soc.* **1997**, *144*, 1609.

to the lower covalence of the M-O bonds in these polyanion structures, the Fe^{3+}/Fe^{2+} and V^{4+}/V^{3+} redox couples lie at much more useful potentials vs Li⁺/Li than in the simple oxides.²⁻⁹

Electrochemical insertion of Li⁺ into Li₃Fe₂(PO₄)₃ leads to Li₅Fe₂(PO₄)₃ (Fe³⁺/Fe²⁺ couple),^{2,8} whereas electrochemical extraction of Li^+ from $Li_3V_2(PO_4)_3$ leads to $LiV_2(PO_4)_3$ (V⁴⁺/V³⁺ couple).^{9,10} These $Li_3M_2(PO_4)_3$ (M = Fe, V) compositions adopt two distinct crystallographic forms, denoted A-Li₃M₂(PO₄)₃ (monoclinic, $P2_1/n$) and B-Li₃M₂(PO₄)₃ (rhombohedral R3, NASI-CON) that differ by the way the $M_2(PO_4)_3$ "lantern units" are connected. The "A" form is the stable form, prepared via solid-state reaction, whereas the "B" form is obtained through ion exchange from the NASICON sodium analogues $Na_3M_2(PO_4)_3$ (M = Fe, V).^{11–13} The A and B forms show distinct electrochemical behavior.^{8,9}

- (9) Gaubicher, J.; Wurm, C.; Goward, G.; Masquelier, C.; Nazar, L. *Chem. Mater.* **2000**, *12*, 3240.
- (10) Gopalakrishnan, J.; Kasthuri-Rangan, K. Chem. Mater. 1992, 4. 745.
- (11) Pintard-Scrépel, M.; d'Yvoire, F.; Rémy, F. C. R. Acad. Sci. 1978, 286, 381.
- (12) d'Yvoire, F.; Pintard-Scrépel, M.; Bretey, E.; De la Rochère,
 M. Solid State Ionics 1983, *9*(10, 851.

(13) d'Yvoire, F.; Pintard-Scrépel, M.; Bretey, E. Solid State Ionics 1986. 18/19. 502.

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[‡] CEA/Saclay.

[§] Université Picardie Jules Verne.

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⁽⁷⁾ Padhi, A. K.; Nanjundaswamy, K. S.; Masquelier, C.; Goodenough, J. B. J. Electrochem. Soc. 1997, 144, 2581.

⁽⁸⁾ Masquelier, C.; Padhi, A. K.; Nanjundaswamy, K. S.; Good-enough, J. B. J. Solid State Chem. 1998, 135, 228.

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A-LFP		$Fe_2(SO_4)_3$	B-I	_FP	Na ₃ Fe ₂ (PO ₄) ₃		
compound	our data	ref 16	$P2_1/n^{15}$	our data	ref 23	ref 18	$R\overline{3}c^{22}$
<i>T</i> _N (K)	25	28	${\sim}30$	23	27	47	47
$\theta_{\rm P}$ (K)	-55	-55	-82	-50	-51(3)	-85	-76
μ_{eff} (μ_{B})	5.89(1)	5.4	5.9	5.77(1)	5.94(3)	4.26	4.5

Our interest lies also in determining the structural factors that have an impact on the electrochemical behavior of these materials. A detailed structural study of the metastable, rhombohedral, form of Li₃Fe₂(PO₄)₃ [noted B-LFP] was communicated recently.¹⁴ As part of our structural synchrotron X-ray and neutron diffraction works on these compositions, it was interesting also to investigate their magnetic structures that were still unknown. Only a few magnetic structural studies have been published so far, on isotypic compositions such as the stable monoclinic "A" form, adopted for instance by $Fe_2(SO_4)_3^{5-17}$ or on the stable "B" form adopted by $Na_3Fe_2(PO_4)_3^{18-22}$ In the course of our study, the magnetic behavior of B-Li₃Fe₂(PO₄)₃ was also reported.²³

Experimental Section

The monoclinic (A-LFP), and rhombohedral (B-LFP) forms of Li₃Fe₂(PO₄)₃ were prepared as described.^{8,12,13} A-Li₃Fe₂(PO₄)₃ was obtained in pure form after a prolonged final solid-state reaction in a Pt crucible at 930 °C from a stoichiometric starting mixture of Li₂CO₃, Fe₂O₃, and NH₄H₂PO₄. B-Li₃Fe₂-(PO₄)₃ was prepared by three successive ion exchanges (1 day each) from Na₃Fe₂(PO₄)₃ in a renewed concentrated aqueous solution of LiNO₃ (Li_{solution}/Na_{solid} > 10). The phase purity was carefully controlled by chemical analysis and X-ray diffraction on a Philips diffractometer (Cu Ka radiation) equipped with a back monochromator, as well as by high-resolution Synchrotron X-ray diffraction on the WD4Č wiggler beamline of the DCI ring of LURE (Laboratoire pour l'Utilisation du Rayonnement Electromagnetique). Neutron diffraction experiments at 300 K and 1.5 K were performed at the Orphée reactor at the Laboratoire Léon Brillouin in Saclay, France. The neutron powder diffractometer G4.2 ($\lambda = 2.3433$ Å) allowed a precise determination of the lattice parameters at room temperature and the resolution of the magnetic structures at 1.5 K. The high-resolution neutron powder diffractometer 3T2 was used for data collection with high direct space resolution ($\lambda = 1.2251$ Å, $Q_{\text{max}} = 9.2$ Å⁻¹), which allowed a precise determination of the nuclear crystal structures at room temperature. The program FullProf²⁴ was used for crystal structure refinements by using the Rietveld method.²⁵

Results and Discussion

A. Magnetic Properties of A-Li₃Fe₂(PO₄)₃ and B-Li₃Fe₂(PO₄)₃. Magnetic and spectroscopic properties of A- and B-LFP have been published recently by two different groups.^{16,23} In complete agreement with these reports, A- and B-Li₃Fe₂(PO₄)₃ demonstrate antiferromagnetic behavior and the values of magnetic moments, paramagnetic Curie temperature (determined from a Curie-Weiss behavior >100 K), and Néel temperature are reported in Table 1. Values found in the literature for similar compounds are also reported for comparison. Both A- and B-LFP samples exhibit Curie-Weiss behavior above 100 K, leading to a magnetic moment of 5.89(1) μ_B (A-LFP) and 5.77(3) μ_B (B-LFP), consistent with the theoretical value of a free high spin Fe^{3+} in a weak crystal field (5.92 $\mu_{\rm B}$). The value obtained for the A-LFP is slightly higher than the relatively low value (5.4 $\mu_{\rm B}$) reported by Goñi et al.¹⁶ The negative paramagnetic Curie temperatures (-50 K and -55 K for Aand B-LFP, respectively) indicate a global antiferromagnetic behavior at low temperatures and is in very good agreement with other reports.¹⁶ The value of the Néel temperature T_N is very close in both samples, about 25 K, also consistent with values already published.^{16,23} These values were used to plan the neutron diffraction experiments. Macroscopic magnetic measurements in A- and B-LFP indicate the existence of possible ferrimagnetism in both compounds.^{16,23} The observed antiferromagnetic ordering is caused by exchange interactions that we identify after the analysis of the magnetic structures.

B. Nuclear and Magnetic Structures of A-Li₃Fe₂-(PO₄)₃ (Monoclinic, P2₁/n). The stable form of Li₃Fe₂- $(PO_4)_3$, A-LFP, is monoclinic at room temperature (α form, space group $P2_1/n$ and transforms on heating to orthorhombic symmetry (γ -form, space group *Pcan*) because of the disappearance of lithium ion ordering in the three-dimensional framework.²⁶ We have refined the crystal structure of the α -form with neutron powder diffraction, at room temperature and at 1.5 K where antiferromagnetic ordering occurs on the iron sublattice.

We first checked the sample quality with conventional X-ray diffraction (Cu K α) at room temperature. All the peaks are indexed in the monoclinic space group $P2_1/n$. The monoclinic distortion is very small [$\gamma = 90.523(1)^\circ$; pseudo-orthorhombic symmetry²⁶] but high-resolution Synchrotron radiation at LURE ($\lambda = 0.9616$ Å), allowed a clear separation of the Bragg peaks (Figure 1). The lattice parameters obtained from Synchrotron X-ray diffraction are in very good agreement with those already published and with those determined from powder neutron diffraction at room temperature (λ =

⁽¹⁴⁾ Masquelier, C.; Wurm, C.; Rodriguez-Carvajal, J.; Gaubicher, J.; Nazar, L. F. Chem. Mater 2000, 12, 522.

⁽¹⁵⁾ Long, G. J.; Longworht, G.; Battle, P.; Cheetham, A. K.; Thundathil, R. V.; Beveridge, D. Inorg. Chem. **1979**, 18, 624. (16) Goñi, A.; Lezama, L.; Moreno, N. O.; Fournès, L.; Olazcuaga,

R.; Barberis, G. E.; Rojo, T. Chem. Mater. 2000, 12(1), 62.

⁽¹⁷⁾ Mesa, J. L.; Goñi, A.; Brandl, A. L.; Moreno, N. O.; Barberis, G. E.; Rojo, T. J. Mater. Chem. 2000, 10, 2779.

⁽¹⁸⁾ Beltran Porter, D.; Olazcuaga, R.; Delmas, C.; Cherkaoui, F.; Brochu, R.; Le Flem, G. Rev. Chim. Minér. 1980, 17, 458. (19) Fanjat, N.; Soubeyroux, J. L. J. Magn. Magn. Mater. 1992,

^{104-107, 933.}

⁽²⁰⁾ Fanjat, N.; Barj, M.; Briat, B.; Schaerpf, O.; Lucazeau, G. J. Phys. Chem. Solids 1993, 54(11), 1515. (21) Fanjat, N.; Lucazeau, G. J. Phys. Chem. Solids 1993, 54(2),

^{187.} (22) Greaves, C.; Slater, P. R.; Slaski, M.; Muirhead, C. M. Physica

B 1994, 194–196, 199.

⁽²³⁾ Andersson, A. S.; Kalska, B.; Jonsson, P.; Haggstrom, L.; Nordblad, P.; Tellgren, R.; Thomas, J. O. J. Mater. Chem. 2000, 10, 2542

⁽²⁴⁾ Rodriguez-Carvajal, J. Physica B 1993, 192, 55. http:// www-llb.cea.fr/fullweb/powder.htm.

⁽²⁵⁾ Rietveld, H. M. J. Appl. Crystallogr. 1969, 2, 65.

⁽²⁶⁾ Bykov, A. B.; Chirkin, A. P.; Demyanets, L. N.; Doronin, S. N.; Genkina, E. A.; Ivanov-Shits, A. K.; Kondratyuk, I. P.; Maksimov, B. A.; Melnikov, O. K.; Muradyan, L. N.; Simonov, V. I.; Timofeeva, V. A. Solid State Ionics 1990, 38, 31.



Figure 1. Comparison between Synchrotron (a) and conventional (b) X-ray diffraction patterns for A-LFP, at room temperature (α -form).

Table 2. Unit Cell Parameters of A-LFP Determined from Neutron Diffraction at 300 K (α -Form), Compared with Those Reported in Ref 26 from Single-Crystal X-ray Diffraction at 293 K

A-Li ₃ Fe ₂ (PO ₄) ₃ , α-form	our data neutron powder diffraction ^a	from Bykov et al. ²⁶ single-crystal X-ray diffraction
temperature (T)	300 K	293 K
wavelength (λ)	2.3433 Å	0.7093 Å
space group	P2 ₁ /n	P2 ₁ /n
a	8.5706(2) Å	8.562(2) Å
b	12.0170(2) Å	12.005(3) Å
c	8.6162(2) Å	8.612(2) Å
γ	90.523(1)	90.51(2)

 a $R_{\rm p},\, 6.44;\, R_{\rm wp},\, 6.90;\, R_{\rm exp},\, 5.00;\, \chi^{2},\, 1.91;\, {\rm Bragg}\,R$ factor, 3.76; N- P + C, 1488; total number of "independent" reflections, 566.

2.3433 Å) (Table 2). Rietveld refinements were performed with the atomic coordinates previously determined by Maksimov from single-crystal X-ray diffraction.²⁷ Refinement of all the atomic coordinates and of the thermal parameters lead to very satisfactory reliability factors listed in Table 2. The list of fractional coordinates is given in Table 3, where the notation "s" [for example Fe(1s)] refers to the (1/2 + x, 1/2 - y, z)pseudosymmetry element. The experimental, calculated, and difference profiles are given in Figure 2. There are two independent crystallographic sites for iron and three for phosphorus. The average Fe-O (2.006 and 2.018 Å) and P-O (1.530, 1.536, and 1.526 Å) bond lengths are characteristic of Fe³⁺ in octahedral coordination and P⁵⁺ in tetrahedral coordination, respectively. As described by Bykov et al.,²⁶ the lithium ions fully occupy three distinct crystallographic sites, two of which are in 5-fold coordination [Li(2) and Li(3)] whereas Li(1) is in 4-fold coordination.

The neutron diffraction pattern recorded when A-Li₃-Fe₂(PO₄)₃ is cooled below $T_{\rm N}$ shows the appearance of some extra reflections. In addition, the intensity of some peaks already present at high temperature is increased (Figure 3). One very intense reflection is clearly observed at $2\theta = 19.3^{\circ}$, and some other reflections are

Table 3. Structural Parameters of A-LFP Determined from Neutron Diffraction at 300 K, α -Form ($P2_1/n$); Atomic Positions of Magnetic Ions in the Unit Cell and Their Magnetic Moment (μ_B) at 1.5 K

	-			
atom	X	У	Ζ	$B_{\rm iso}$ (Å ²)
Fe(1)	0.2462(5)	0.1079(3)	0.4619(4)	0.26(7)
Fe(1s)	0.7541(5)	0.3950(3)	0.4707(4)	0.22(7)
P(1)	0.1030(8)	0.1484(5)	0.1069(8)	0.9(2)
P(1s)	0.6039(9)	0.3495(5)	0.1160(8)	0.5(1)
P(2)	0.0359(7)	0.4932(5)	0.251(1)	1.0(1)
O(1)	0.4292(7)	0.3309(5)	0.0885(7)	0.9(1)
O(1s)	0.9237(7)	0.1497(4)	0.1151(7)	0.6(1)
O(2)	0.3527(7)	0.2620(5)	0.4806(7)	0.6(1)
O(2s)	0.8022(7)	0.2193(4)	0.4972(6)	0.3(1)
O(3)	0.1669(8)	0.0389(5)	0.0580(7)	1.4(1)
O(3s)	0.6455(7)	0.4713(5)	0.0920(7)	0.8(1)
O(4)	0.4506(7)	0.0689(4)	0.3678(8)	0.4(1)
O(4s)	0.9273(7)	0.4037(5)	0.3132(7)	0.6(1)
O(5)	0.1709(6)	0.4317(4)	0.1710(7)	0.4(1)
O(5s)	0.5979(7)	0.0701(4)	0.1273(8)	0.6(1)
O(6)	0.1635(6)	0.1868(4)	0.2636(7)	0.3(1)
O(6s)	0.6374(7)	0.3168(5)	0.2868(8)	1.0(1)
Li(1)	0.288(2)	0.320(2)	0.275(2)	0.05(35)
Li(2)	0.571(2)	0.202(2)	0.419(2)	1.4(5)
Li(3)	0.901(3)	0.243(2)	0.291(3)	5.2(7)

Magnetic Moments (μ_B) at 1.5 K

	-					
atom	X	У	Z	M_X	M_Y	M_Z
Fe(1-1)	X	у	Ζ	4.7(3)	0	0
Fe(1-2)	-x + 1/2	-y + 1/2	z + 1/2	4.7(3)	0	0
Fe(1-3)	-x	-y	-z	4.7(3)	0	0
Fe(1-4)	x + 1/2	y + 1/2	-z + 1/2	4.7(3)	0	0
Fe(1s-1)	X	y	Z	-4.0(3)	0	0
Fe(1s-2)	-x + 1/2	-y + 1/2	z + 1/2	-4.0(3)	0	0
Fe(1s-3)	-x	-y	-z	-4.0(3)	0	0
Fe(1s-4)	x + 1/2	y + 1/2	-z + 1/2	-4.0(3)	0	0

more intense at 1.5 K than at 40 K. All these reflections can be indexed with a propagation vector $\mathbf{k} = (0,0,0)$, which indicates that the magnetic unit cell is the same as the nuclear unit cell described above and, consequently, the space group remains $P2_1/n$, with lattice parameters a = 8.5599(2) Å, b = 11.9965(3) Å, c = 8.6096(2) Å, and $\gamma = 90.559(2)^\circ$. The eight magnetic Fe³⁺ ions of the cell are distributed in two different crystallographic sites, at general position *4e*, with $x \approx 0.246$, $y \approx 0.108$, $z \approx 0.462$ for Fe(1) and $x \approx 0.754$, $y \approx 0.395$, $z \approx 0.471$ for Fe(1s).

⁽²⁷⁾ Maksimov, B. A.; Muradyan, L. A.; Genkina, E. A.; Simonov, V. I. Sov. Phys. Dokl. 1986, 31(5), 370.



Figure 2. Calculated and experimental neutron diffraction patterns of A-LFP at 300 K ($\lambda = 2.34$ Å).

We labeled the four iron atoms generated by Fe(1) in position 4e as Fe(1–1), Fe(1–2), Fe(1–3), and Fe(1–4). The different possibilities of magnetic configurations were investigated by using Bertaut's symmetry analysis method,^{28,29} which allows the determination of the symmetry constraints between each magnetic moment of Fe³⁺ belonging to the same general crystallographic position. The representation of the propagation vector group can be decomposed upon four irreducible representations, each with three basis vectors, $\Gamma = 3$ ($\Gamma_1 +$ $\Gamma_2 + \Gamma_3 + \Gamma_4$), leading to four possible spin configurations:

$$\Gamma_{1}: \mathbf{G}^{X} = \mathbf{S}_{1}^{X} - \mathbf{S}_{2}^{X} + \mathbf{S}_{3}^{X} - \mathbf{S}_{4}^{X}; \mathbf{G}^{Y} = \mathbf{S}_{1}^{Y} - \mathbf{S}_{2}^{Y} + \mathbf{S}_{3}^{Y} - \mathbf{S}_{4}^{Y}; \mathbf{F}^{Z} = \mathbf{S}_{1}^{Z} + \mathbf{S}_{2}^{Z} + \mathbf{S}_{3}^{Z} + \mathbf{S}_{4}^{Z}$$

$$\Gamma_{2}: \mathbf{A}^{X} = \mathbf{S}_{1}^{X} - \mathbf{S}_{2}^{X} - \mathbf{S}_{3}^{X} + \mathbf{S}_{4}^{X}; \mathbf{A}^{Y} = \\ \mathbf{S}_{1}^{Y} - \mathbf{S}_{2}^{Y} - \mathbf{S}_{3}^{Y} + \mathbf{S}_{4}^{Y}; \mathbf{C}^{Z} = \mathbf{S}_{1}^{Z} + \mathbf{S}_{2}^{Z} - \mathbf{S}_{3}^{Z} - \mathbf{S}_{4}^{Z}$$

$$\Gamma_{3}: F^{X} = S_{1}^{X} + S_{2}^{X} + S_{3}^{X} + S_{4}^{X}; F^{Y} = S_{1}^{Y} + S_{2}^{Y} + S_{3}^{Y} + S_{4}^{Y}; G^{Z} = S_{1}^{Z} - S_{2}^{Z} + S_{3}^{Z} - S_{4}^{Z}$$

$$\Gamma_{4}: C^{X} = S_{1}^{X} + S_{2}^{X} - S_{3}^{X} - S_{4}^{X}; C^{Y} = S_{1}^{Y} + S_{2}^{Y} - S_{3}^{Y} - S_{4}^{Y}; A^{Z} = S_{1}^{Z} - S_{2}^{Z} - S_{3}^{Z} + S_{4}^{Z},$$

where S_i^X is the component along *x* of the magnetic moment of atom *i* for a general position Fe(j) (j = 1 or 1s in our case). For example, the representation Γ_1 corresponds to a ferromagnetic coupling of the four moments in the *z* direction (F^Z), whereas in the *x* and *y* directions (G^X and G^Y), the moments are coupled antiferromagnetically between Fe(j-1) and Fe(j-2) and between Fe(j-3) and Fe(j-4), and ferromagnetically between Fe(j-1) and Fe(j-3).

We tried all these magnetic models by least-squares refinements and Monte Carlo techniques. Refinement of independent magnetic moments for Fe(1) and Fe(1s) led to an antiferromagnetic coupling along the *x* axis. Of all the possibilities, the collinear solution $F^X = S_1^X$



Figure 3. Neutron diffraction patterns of A-LFP showing the appearance of magnetic peaks at low temperature.

 $+ S_2^X + S_3^X + S_4^X(\Gamma_3)$, with opposite magnetic moments for Fe(1) and Fe(1s), gives the best agreement between observed and calculated patterns. No component of the magnetic moment exists along the y and z directions. We first tried the refinement with a same magnetic moment amplitude for Fe(1) and Fe(1s), and at the last stage of the refinement, we allowed the magnetic moments of Fe(1) and Fe(1s) to be completely independent. The refinement is shown in Figure 4, and the magnetic final Bragg *R* factor is 5.34%. For Fe^{3+} in highspin configuration, the expected moment is 5 $\mu_{\rm B}$, but we observed slightly smaller values, perhaps because of a combined effect between the covalent character of the Fe-O bonds and the zero-point spin fluctuation of antiferromagnetism. The refined values for the magnetic moment are 4.7(3) μ_B for Fe(1) and 4.0(3) μ_B for Fe(1s), respectively. The magnetic moments for the eight iron magnetic ions of the unit cell are reported in Table 3. The difference observed between the magnetic moments of Fe(1) and Fe(1s) is in total agreement with macroscopic magnetic data at low temperature, indicating a ferrimagnetic structure for this compound.^{16,17} The structure we found by neutron diffraction confirms experimentally the prediction of Goñi et al. from electron spin resonance spectra, the magnetic interaction in each sublattice being ferromagnetic and antiferromagneti-

⁽²⁸⁾ Bertaut, E. F. J. Phys. 1971, 32, C1.

⁽²⁹⁾ Rossat-Mignot, J. Magnetic Structures and Neutron Diffraction; Academic Press: New York, 1987.



Figure 4. Observed (obs, circles) versus calculated (calc, continuous line) neutron powder diffraction patterns of A-LFP ($\lambda = 2.34$ Å) at 1.5 K. The positions of the Bragg reflections are represented by vertical bars (first row, nuclear; second row, magnetic). The difference (obs – calc) pattern is displayed at the bottom of the figure. Arrows indicate the most intense magnetic reflections.



Figure 5. Representation of the magnetic structure for A-LFP. Filled and open circles represent opposite directions of the magnetic moment along [001].

cally coupled between the two Fe(1) and Fe(1s) sublattices. Figure 5 represents the magnetic structure found by neutron diffraction. The filled circles correspond to a negative magnetic moment of 4.0(3) $\mu_{\rm B}$ in the x direction, whereas the open circles correspond to a positive moment of 4.7(3) $\mu_{\rm B}$ in the *x* direction. In each 'lantern unit', the two Fe(1) and Fe(1s) are antiparallel, whereas two neighboring lantern units are parallel. This result can be compared with the magnetic structure of the related iron sulfate $Fe_2(SO_4)_3$.¹⁵ This Fe^{3+} sulfate presents the same nuclear structure (space group $P2_1/n$) for the framework, and the arrangement of the magnetic moments is very similar to what we observe in A-Li₃Fe₂(PO₄)₃; the orientation of magnetic moments between crystallographically equivalent iron atoms is parallel, and Fe(1) and Fe(1s) atoms are antiparallel lying in the (a, b) plane. The magnetic moment, constrained to be equal for Fe(1) and Fe(1s), was refined

to 4.52(8) $\mu_{\rm B}$. In A-Li₃Fe₂(PO₄)₃, introducing a magnetic component along the *y* axis (according to the Γ_3 representation) led to a refined value for the *y* component very close to 0 $\mu_{\rm B}$, so, within the experimental error, the magnetic structure of A-Li₃Fe₂(PO₄)₃ can be described by (F_X⁽¹⁾, - F_X⁽²⁾).

Analysis and Discussion of the Magnetic Structure. Magnetic Phase Diagram. The relative impacts of the different exchange interactions within this crystal structure type on the observed magnetic structure were closely examined. Because of the complexity of the topology, we had to treat the problem numerically by using the two computer programs, SIMBO and ENER-MAG, which are briefly described in one of our previous articles.³⁰ The symmetry of the structure is so low that an analysis of the exchange paths led to 18 different exchange interactions, all of them being of the supersuper exchange type; two anions are involved in the path. These interactions only differ by some changes in distances or in $(Fe-O_i-O_j)$ angles, (O_i-O_j-Fe) angles and (Fe–O_i–O_i–Fe) torsion angles. We chose to simplify this problem by using the highest-symmetry structure, that is the orthorhombic space group Pcan (#60). The number of atoms is then divided by 2, because the pseudo-symmetry element (1/2 + x, 1/2 - y, z) observed in the nuclear structure is taken into account. P(2) lies in special position 4c. The atomic positions we used for each iron and oxygen atom are the mean values of the previous two crystallographic positions in the space group $P2_1/n$. Note that this average structure corresponds to the high-temperature form of A-LFP (γ -form, >530 K²⁶). In this space group, the number of independent exchange interactions found by SIMBO is 9, but this number can be reduced to 5 by a careful analysis of the exchange paths; we assume that the exchange integrals will have the same value if the topology is similar. The Fe–O distances are almost all identical, and we compared the $(Fe-O_i-O_i)$ and (O_i-O_i-Fe) angles and torsion angles to distinguish between the different paths. Then, apart from J_1 which connects the iron atoms belonging to a same 'lantern unit', the exchange paths can be grouped by 2. Table 4 reports the five resulting exchange interactions that should be considered to study the problem and the related geometrical information. Figure 6 represents these interactions (here the atoms are labeled in the *Pcan* space group, i.e., the notation 's' is omitted, except for Fe for reasons of clarity). J_1 is the coupling between two iron atoms that belong to the same lantern unit; the magnetic moments are aligned perpendicular to the general direction of the lantern units and are coupled antiferromagnetically. These two Fe³⁺ are coupled via three different super-super exchange paths involving O(3) and O(6), O(4) and O(4), and O(6) and O(3), corresponding to the three PO_4 tetrahedra that link the two $[Fe(1)O_6]$ and $[Fe(1s)O_6]$ octahedra (Figure 6). This antiferromagnetic exchange may be the most intense because the Fe-Fe distance is the smallest in the structure (4.517 Å), and it governs the global antiferromagnetic properties of the material. The four other magnetic couplings, J_2 , J_3 , J_4 , and J_5 , occur via two double super-super

⁽³⁰⁾ El Khayati, N.; Cherkaoui El Moursli, R.; Rodriguez-Carvajal, J.; Andre, G.; Blanchard, N.; Bouree, F.; Collin, G.; Roisnel, T. *Eur. J. Phys. B* **2001**, *22*(4), 429.

Table 4. List of Effective Exchange Interactions Considered between Iron Atoms and Related Super-Super Exchange Paths, Bond Lengths, and Angles for A-LFP^a

interaction	path	d1	d2	d3	α1	α2	γ	d
J_1	Fe(1)-O(3)-O(6)-Fe(1s)	1.90	2.50	2.08	141.2	102.0	164.4	4.52
	Fe(1)-O(4)-O(4)-Fe(1s)	1.97	2.52	1.97	117.1	117.1	134.7	4.52
	Fe(1)-O(6)-O(3)-Fe(1s)	2.08	2.50	1.90	102.0	141.2	164.4	4.52
J_2	Fe(1)-O(4)-O(5)-Fe(1)	1.97	2.47	1.91	130.0	123.4	123.1	5.02
	Fe(1)-O(5)-O(4)-Fe(1)	1.91	2.47	1.97	123.4	130.0	123.1	5.02
	Fe(1)-O(1)-O(3)-Fe(1)	2.01	2.49	1.90	110.4	152.7	152.3	5.02
	Fe(1)-O(3)-O(1)-Fe(1)	1.90	2.49	2.01	152.7	110.4	152.3	5.02
J_3	Fe(1)-O(1)-O(2)-Fe(1s)	2.01	2.46	2.10	121.3	128.7	76.4	5.49
	Fe(1)-O(2)-O(1)-Fe(1s)	2.10	2.46	2.01	128.7	121.3	76.4	5.49
	Fe(1)-O(6)-O(2)-Fe(1)	2.08	2.53	2.10	146.6	108.1	77.5	5.51
	Fe(1)-O(2)-O(6)-Fe(1)	2.10	2.53	2.08	108.1	146.6	77.5	5.51
J_4	Fe(1)-O(5)-O(5)-Fe(1s)	1.91	2.40	1.91	139.2	139.2	104.5	5.51
	Fe(1)-O(3)-O(2)-Fe(1s)	1.90	2.46	2.10	139.7	160.8	99.5	6.03
	Fe(1)-O(2)-O(3)-Fe(1s)	2.10	2.46	1.90	160.8	139.7	99.5	6.03
J_5	Fe(1)-O(4)-O(5)-Fe(1s)	1.97	2.35	1.91	168.4	158.1	150.1	6.07
	Fe(1)-O(5)-O(4)-Fe(1s)	1.91	2.35	1.97	158.1	168.4	150.1	6.07
	Fe(1)-O(6)-O(1)-Fe(1s)	2.08	2.45	2.01	138.3	168.8	147.2	6.07
	Fe(1)-O(1)-O(6)-Fe(1s)	2.01	2.45	2.08	168.8	138.3	147.2	6.07

^a The pseudo-orthorhombic symmetry (space group Pcan) has been considered, for which 's' atoms are deduced from their corresponding analogues by the symmetry element (1/2 + x, 1/2 - y, z). d1 = $d(\text{Fe}-\text{O}_i)$; d2 = $d(\text{O}_i-\text{O}_i)$; d3 = $d(\text{O}_i-\text{Fe})$; a1 = (Fe-O_i-O_i); a2 = $d(\text{O}_i-\text{O}_i)$; d3 = $d(\text{O}_i-\text{Fe})$; a1 = (Fe-O_i-O_i); a2 = $d(\text{O}_i-\text{O}_i)$; d3 = $d(\text{O}_i-\text{Fe})$; a1 = (Fe-O_i-O_i); a2 = $d(\text{O}_i-\text{O}_i)$; d3 = $d(\text{O}_i-\text{Fe})$; a1 = (Fe-O_i-O_i); a2 = $d(\text{O}_i-\text{O}_i)$; d3 = $d(\text{O}_i-\text{Fe})$; a1 = (Fe-O_i-O_i); a2 = $d(\text{O}_i-\text{O}_i)$; d3 = $d(\text{O}_i-\text{Fe})$; a1 = (Fe-O_i-O_i); a2 = $d(\text{O}_i-\text{O}_i)$; a2 = $d(\text{O}_i-\text{O}_i)$; a2 = $d(\text{O}_i-\text{O}_i)$; a3 = $d(\text{O}_i-\text{Fe})$; a1 = (Fe-O_i-O_i); a2 = $d(\text{O}_i-\text{O}_i)$; a2 = $d(\text{O}_i-\text{O}_i)$; a3 = $d(\text{O}_i-\text{O}_i)$; a2 = $d(\text{O}_i-\text{O}_i)$; a3 = $d(\text{O}_i-\text{Fe})$; a1 = (Fe-O_i-O_i); a2 = $d(\text{O}_i-\text{O}_i)$; a2 = $d(\text{O}_i-\text{O}_i)$; a3 = $d(\text{O}_i-\text{O}_i)$; a2 = $d(\text{O}_i-\text{O}_i)$; a3 = $d(\text{O}_i-\text{O}_i)$; a4 = $d(\text{O}_i-\text{O}_i)$; a5 = d(\text{O}_i-\text{O}_i); a5 = $d(\text{O}_i-\text{O}_i)$; a6 = $d(\text{O}_i-\text{O}_i)$; a6 = $d(\text{O}_i-\text{O}_i)$; a6 = $d(\text{O}_i-\text{O}_i)$; a6 = $d(\text{O}_i-\text{O}_i)$; a7 = d(\text{O}_i-\text{O}_i); a7 = d(\text{O}_i-\text{O}_i); a7 = d(\text{O}_i-\text{O}_i); a7 = d(\text{O}_i-\text{O}_i); a8 = d(\text{ $(O_i - O_i - Fe)$; γ = torsion angle; d = d(Fe - Fe).



Figure 6. Representation of the five exchange paths between the iron atoms that must be considered in A-LFP.

exchange paths involving two oxygen atoms. J_2 links two Fe(1) via O(4) and O(5) in the (a, c) plane, and via O(1) and O(3) in the (a, b) plane. J_3 and J_4 connect two adjacent lantern units. We tried to evaluate the relative strengths and signs of these four exchange interactions that give the observed magnetic structure as the ground state. For that, numerical phase diagrams were generated by the program ENERMAG.³⁰

The problem of the magnetic ground state of a system of classical spins connected by isotropic exchange interactions was considered 40 years ago by several authors. $^{31-34}$ Here we follow the discussion summarized by Fraiser.³⁴ The first ordered state can be obtained from the resolution of an eigenvalue problem in which the matrix is the Fourier transform of the exchange interactions. In our case we do not have any new magnetic transition below $T_{\rm N}$, so that the first ordered magnetic state is the ground state. We use the method discussed in refs 31 and 34 to evaluate the conditions

satisfied by the exchange integrals to have the propagation vector $\mathbf{k} = (0,0,0)$ and the observed spin arrangement (irreducible representation Γ_3), ($F_{X}^{(1)}$, $-F_{X}^{(2)}$) = (++++; ---), as the ground state.

The energy, lowest eigenvalue of the matrix $\xi(\mathbf{k}, \{J_{ii}\})$, as a function of the exchange integrals and $\mathbf{k} = (X, Y, Y)$ Z), can be obtained only numerically. The vector ${\bf k}$ minimizing $\lambda(\mathbf{k}, \{J_{ii}\})$ for a given set of $\{J_{ii}\}$ is the propagation vector of the magnetic structure, and the spin configuration is obtained from the corresponding eigenvector. For the cases in which $\mathbf{k} = 1/2\mathbf{H}$, **H** being a reciprocal lattice vector, the eigenvectors are all real and the sequence of signs of the eigenvector components corresponding to the lowest eigenvalue gives the spin configuration corresponding to the ground state.

A careful analysis of the five exchange integrals led us to make $J_3 = 0$, because this path presents angles that do not favor exchange interactions; the torsion angle is close to 80° and we expect this interaction to be negligible compared with others. This also reduces the number of integrals to consider to four, which is reasonable for plotting a phase diagram. To study the problem with ENERMAG, we varied the values of all the exchange interactions J_i (i = 2, 4, 5) taking J_1 as a reference value ($J_1 = +100 \text{ or } J_1 = -100$) and $J_3 = 0$, in the interval [-100, 100]. The **k** vectors were varied inside the Brillouin zone and in special points. An auxiliary program takes the output of ENERMAG and plots a high-dimensional phase diagram with the exchange interactions as Cartesian axes. The different regions correspond to different magnetic structures. We have numbered the eight different kinds of collinear magnetic structure, with $\mathbf{k} = (0,0,0)$, found by the program, and numbered as '9' the regions where there is no classical magnetic ordering or where the magnetic structure is incommensurate because of frustration effects. The sign sequence characterizing the eight collinear structures is reported in Table 5. The observed magnetic structure of A-LFP is given by the sequence $(F_X^{(1)}, -F_X^{(2)}) = (++++; ---)$ and is numbered as'1'

In Figure 7 we have represented two-dimensional maps of the different regions for representative cases.

⁽³¹⁾ Yoshimori, A. J. Phys. Soc. Jpn. 1959, 14, 807.

⁽³²⁾ Villain, J. J. Phys. Chem. Solids 1959, 11, 303.
(33) Lyons, D. H.; Kaplan, T. Phys. Rev. 1960, 120, 1580.
(34) Fraiser, M. J. Phys. Rev. 1961, 123, 2003.

Table 5. Label of the Magnetic Structures and Sign Sequences of the Magnetic Moments Corresponding to Atoms Fe(1-1), Fe(1-2), Fe(1-3), Fe(1-4), Fe(1s-1), Fe(1s-2), Fe(1-3), and Fe(1s-4) Characterizing the Eight Possible Magnetic Structures for k = (0,0,0) for A-LFP^a

	sign sequences of collinear magnetic moments								
structure	M _{Fe(1-1)}	M _{Fe(1-2)}	M _{Fe(1-3)}	M _{Fe(1-4)}	$M_{Fe(1s-1)}$	M _{Fe(1s-2)}	M _{Fe(1s-3)}	$M_{Fe(1s-4)}$	
1. $\mathbf{k} = (0,0,0)$	+	+	+	+	_	_	_	_	
2. $\mathbf{k} = (0,0,0)$	+	_	_	+	+	_	_	+	
3. $\mathbf{k} = (0,0,0)$	+	+	_	_	_	_	+	+	
4. $\mathbf{k} = (0,0,0)$	+	_	+	_	+	_	+	_	
5. $\mathbf{k} = (0,0,0)$	+	+	_	_	+	+	_	_	
6. $\mathbf{k} = (0,0,0)$	+	_	_	+	_	+	+	_	
7. $\mathbf{k} = (0,0,0)$	+	+	+	+	+	+	+	+	
8. $\mathbf{k} = (0,0,0)$	+	_	+	_	_	+	_	+	
9			case of i	ncommensurat	e or disordered	structures			

^{*a*} We have numbered as '9' the incommensurate and disordered structure occurring in the magnetic phase diagram because of frustration effects.



Figure 7. Magnetic phase diagram for A-LFP, for $J_1 = -100$. Phases numbered from 1 to 8 correspond to $\mathbf{k} = (0,0,0)$ collinear magnetic structures found by the program ENERMAG. The sequence of signs is described in Table 5. The label '9' corresponds to incommensurate or disordered structures.

An analysis of the boundaries between the regions gives us the conditions that the exchange integrals have to satisfy to give, as the first ordered state, the observed magnetic structure. Only two values of J_1 are considered: +100 and -100, because no fundamental difference is observed on the phase diagrams when $|J_1|$ varies from 0 to +100. Here we have plotted J_4 vs J_2 for different values of J_5 . The eight different collinear structures are all observed, whatever the sign of J_1 . However, their relative importance is different. For example, the ferromagnetic structure $(F_X^{(1)}, F_X^{(2)}) =$ (++++;++++) labeled as '7' is observed in the upper right side of the diagram (J_2 and J_4 positive) when J_5 is positive. This domain is smaller for $J_1 = -100$ than for $J_1 = +100$, but exists. On the opposite, the arrangement (+ + - -; + + - -) labeled as '5', is observed for negative values of J_2 and J_4 , but positive values of J_5 . This structure is obtained for both $J_1 = +100$ and $J_1 = -100$, but is smaller in the former case. Let us concentrate on the magnetic structure we observed from neutron data. It is labeled as '1' and corresponds

to a spin arrangement of $(F_X^{(1)}, -F_X^{(2)}) = (++++; ----)$. The first condition to get the observed magnetic structure as the ground state is J_4 being negative. For $J_1 = -100$, the arrangement '1' is observed if the following conditions are fulfilled:

if $J_5 < 1/_2$, J_1 , $J_4 < 0$, and $J_4 < 2$, $J_2 + 1/_2$, J_1 ; if $|J_5| < 1/_2$, J_1 , then the conditions are $J_2 > 2$, J_5 and $J_4 < 0$;

if $J_5 > -1/2 J_1$, the '2' domain is not observed.

The '1' domain is much smaller for $J_1 = +100$ than for $J_1 = -100$, but the general trend is the same. The conditions are then $J_4 < 0$ and $J_4 < 2J_2 + 1/2J_1$. The domain '1' vanishes very quickly if J_5 increases: for J_5 $> -1/2 J_1$ and above, the domain is not observed. To summarize, the stronger negative J_1 , J_4 , and J_5 , and the stronger positive J_2 , the more important domain '1' is.

C. Nuclear and Magnetic Structures of B-Li₃Fe₂-(PO₄)₃ (Rhombohedral, *R*3). The rhombohedral B-Li₃-Fe₂(PO₄)₃ (B-LFP) compound presents the NASICON

Table 6. Structural Parameters of the B-LFP Compound Determined from Neutron Diffraction, at 300 K ($\lambda = 1.225$

			A)		
atom	X		У	Ζ	<i>B</i> (Ų)
Fe(1)	0	0		0.1461(1)	0.33(5)
Fe(2)	0	0		0.6527(2)	0.36(5)
Р	0.2926(4)	0.0	0014(6)	0.2511(2)	0.39(4)
O(1)	0.1923(6)	-0.0	0106(6)	0.1923(2)	2.64(9)
O(1s)	0.7658(5)	0.9	9132(5)	0.6990(1)	1.42(6)
O(2)	0.2426(5)	-0.1	1995(5)	0.2658(2)	1.50(7)
O(2s)	0.5005(4)	0.8	3795(5)	0.7558(2)	1.08(6)
Li(1)	0.346(2)	0.0	029(2)	0.3808(7)	2.6(2)
	Magne	tic Mor	nents ($\mu_{\rm B}$)) at 1.5 K	
atom	X	У	Z	M_X	M_Z
Fe(1-1)	0	0	Ζ	4.76(3)	0
Fe(1-2)	0	0	-z	4.76(3)	0
Fe(2-1)	0	0	Ζ	-4.76(3)	0
Fe(2-2)	0	0	-z	-4.76(3)	0

^{*a*} Space group $R\overline{3}$, *a* = 8.3165(4) Å, *c* = 22.459(1) Å. *R*_p, 9.94%; *R*_{wp}, 10.1%; *R*_{exp}, 4.88%; χ^2 , 4.32. Atomic positions of magnetic ions in unit cell and their magnetic moment (μ_B) at 1.5 K.

structure with space group $R\overline{3}$; its structure was presented for the first time and in detail in one of our previous articles.¹⁴ Prepared by ion exchange from Na₃Fe₂(PO₄)₃, this compound is based on the same Fe₂- $(PO_4)_3$ lantern units as in A-LFP, but the relative positions of these lantern units are different. Table 6 reports the space group, lattice parameters, and atomic positions of the B-LFP at 300 K, determined from neutron diffraction on the 3T2 diffractometer, starting from the atomic positions of ref 14. One of the most striking features of this structure is that the Li⁺ ions lie in a new 4-fold coordinated site, different from the M(1) and M(2) sites occupied in Na₃Fe₂(PO₄)₃. Note that the same interesting redistribution of lithium ions within the framework occurs on going from B-Na₃V₂- $(PO_4)_3$ to B-Li₃V₂(PO₄)₃.³⁵ We concentrate here on the magnetic structure revealed by neutron diffraction below 25 K, because the neutron diffraction patterns present some new peaks and a higher intensity for some other peaks already present (Figure 8). Although the magnetic properties of the Na analogue $B-Na_3Fe_2(PO_4)_3$ (space group C2/c) has been widely studied, ^{19–21} those of B-LFP have scarcely been mentioned apart from a recent publication that appeared during the course of our study.²³ The results obtained by Andersson et al. on the magnetic structure are slightly different from what we found with our neutron data. We will discuss that point later.

As mentioned by Andersson et al., all the magnetic reflections observed in the neutron diffraction pattern at 1.5 K can be indexed within the nuclear cell, indicating that the propagation vector for the magnetic structure is $\mathbf{k} = (0,0,0)$, as in A-LFP. The magnetic structure



Figure 8. Neutron diffraction patterns of B-LFP showing the appearance of magnetic peaks at low temperature.

was solved by testing the different basis functions of the irreducible representation for $R\overline{3}$ with $\mathbf{k} = (0,0,0)^{.28,29}$ There are two different crystallographic sites for iron in B-LFP, both in the δc Wyckoff position with atomic coordinates of the atoms within a primitive cell (0, 0, z)and (0, 0, -z). The symmetry analysis with $\mathbf{k} = (0,0,0)$ gives a decomposition within six irreducible representations of dimension 1, as $\Gamma = \Gamma_1 + \Gamma_2 + \Gamma_3 + \Gamma_4 + \Gamma_5 +$ Γ_6 . Γ_3 , Γ_4 , Γ_5 , and Γ_6 have complex basis vectors, and are equivalent to two reducible representations Γ_3 and Γ_4 with two real basis vectors. We also have four different representations with basis vectors as follows:

representation	F	'e(1)			Fe(2)	
Γ_1	(0	0	1)	(0	0	1)
Γ_2	(0	0	1)	(0	0	-1)
Γ_{3}'	(1	0	0)	(1	0	0)
Γ_{3}'	(1/2	-1	0)	(1/2	-1	0)
Γ_4'	(1	0	0)	(-1	0	0)
Γ_4'	(-1/2	-1	0)	(1/2	1	0)

The presence of a (001) reflection indicates that there is a magnetic component in the (*a*, *b*) plane. The structure is very well refined with the collinear model $(F_X^{(1)}, -F_X^{(2)})$ corresponding to the Γ_3 ' representation. The values of the magnetic moments of Fe(1) and Fe(2) were constrained to be equal, otherwise the refinement leads to an unrealistic value for one of them (>5 μ_B). The obtained value for Fe(1) and Fe(2) is refined to 4.76(3) μ_B . Note that we cannot distinguish between moments along the *a* or *b* axes with powders in a

Table 7. List of Exchange Interactions and Related Super-Super Exchange Paths, Bond Lengths, and Angles for B-LFP^a

interaction	path	d1	d2	d3	α1	α2	γ	d
J_1	Fe(1)-O(1)-O(1s)-Fe(2)	1.94	2.52	2.00	117.0	118.5	137.1	4.54
	Fe(1)-O(1)-O(1s)-Fe(2)	1.94	2.52	2.00	117.0	118.5	137.1	4.54
	Fe(1)-O(1)-O(1s)-Fe(2)	1.94	2.52	2.00	117.0	118.5	137.1	4.54
J_2	Fe(2)-O(1s)-O(2)-Fe(2)	2.00	2.54	1.99	107.5	125.0	100.4	4.85
	Fe(2)-O(2)-O(1s)-Fe(2)	1.99	2.54	2.00	125.0	107.5	100.4	4.85
J_3	Fe(1)-O(1)-O(2s)-Fe(1)	1.94	2.52	2.08	156.5	98.8	139.9	4.87
	Fe(1)-O(2s)-O(1)-Fe(1)	2.08	2.52	1.94	98.8	156.5	139.9	4.87

 a d1 = $d(Fe-O_{i})$; d2 = $d(O_{i}-O_{j})$; d3 = $d(O_{j}-Fe)$; α 1 = (Fe-O_{i}-O_{j}); α 2 = (O_{i}-O_{j}-Fe); γ = torsion angle; d = d(Fe-Fe).

Table 8. Label of the Magnetic Structures and Sign Sequences of the Magnetic Moments Corresponding to AtomsFe(1-1), Fe(2-1), Fe(2-1), and Fe(2-2) CharacterizIng the Four Possible Collinear Magnetic Structures for k = (0,0,0)for B-LFP

	sign se	sign sequences of collinear magnetic moments			conditions t	o be observed as g	round state
structure	M _{Fe(1-1)}	M _{Fe(1-2)}	M _{Fe(2-1)}	M _{Fe(2-2)}	J_1	J_2	J_3
1. $\mathbf{k} = (0,0,0)$	+	_	_	+	>0	<0	<0
2. $\mathbf{k} = (0,0,0)$	+	+	+	+	>0	>0	>0
3. $\mathbf{k} = (0,0,0)$	+	_	+	_	<0	<0	<0
4. $\mathbf{k} = (0, 0, 0)$	+	+	_	_	<0	>0	>0
5			case of incon	imensurate or dis	ordered structure	S	

^{*a*} We have numbered as '5' the incommensurate and disordered structure occurring in the magnetic phase diagram because of frustration effects. The conditions of existence as ground state are reported in terms of the three super-super exchange interactions described in Table 7.



Figure 9. Observed (obs, circles) versus calculated (calc, continuous line) neutron powder diffraction patterns of B-LFP ($\lambda = 2.34$ Å) at 1.5 K. Position of the Bragg reflections are represented by vertical bars (first row, nuclear; second row, magnetic). The difference (obs – calc) pattern is displayed at the bottom of the figure. Arrows indicate the most intense magnetic reflections.

rhombohedral symmetry. We know, however, that a weak ferrimagnetic component exists from magnetic measurements.²³ This component is probably very weak because we cannot obtain it from neutron diffraction. Andersson et al. suggest a ferromagnetic contribution of 0.7 $\mu_{\rm B}$ along the *z* axis, which would mean that they have a mixture of the Γ_1 and Γ_3 representations. If we try to introduce a weak component along the z axis with our data, the refinement becomes unstable. Table 6 gives the magnetic moments of the magnetic ions within the unit cell. Figure 9 presents the final refinement. (The magnetic reflections are indicated by arrows.) The final magnetic R factor was 7.42%. The magnetic structure obtained is in agreement with what has been observed for the Na analogues¹⁹⁻²¹: the two Fe inside the same lantern unit have an antiparallel orientation, and the orientation between crystallographically equivalent Fe is parallel (Figure 10).

To understand the observed magnetic structure in terms of super-super exchange interactions, the same procedure as for A-LFP was followed. In Table 7 we report the first three super-super-exchange paths connecting two iron atoms. Only three exchange paths have to be considered for this structure. We limited at 5 Å



Figure 10. Representation of the magnetic structure for B-LFP. Filled and open circles represent opposite directions of the magnetic moment in a direction perpendicular to [001].



Figure 11. Representation of the three exchange paths between the iron atoms that must be considered in B-LFP.

the maximum cation-cation distance. Above 5 Å, there is a gap in intercation distance, the first neglected interaction between Fe and Fe being at a distance of at least 5.5 Å. J_1 , J_2 , and J_3 exchange interactions are illustrated in Figure 11. J_1 is the shortest one and connects Fe(1) and Fe(2) in the same lantern unit via three equivalent exchange paths involving O(1) and O(1s). J_2 connects two Fe(2) atoms via O(1s) and O(2). This interaction is perpendicular to the sheets containing the lantern units. J_3 connects two Fe(2) belonging

⁽³⁵⁾ Masquelier, C.; Wurm, C.; Elkaim, E.; Lauriat, J. P., manuscript in preparation.



Figure 12. Magnetic phase diagram for B-LFP. Phases numbered 1, 2, 3, and 4 correspond to $\mathbf{k} = (0,0,0)$ collinear magnetic structures found by the program ENERMAG. The sequence of signs is described in Table 8. The label '5' corresponds to incommensurate or disordered structures.

to two adjacent lantern units, via O(1) and O(2s). As for A-LFP, we calculated the phase diagrams with J_1 , J_2 , and J_3 as Cartesian axes. We have numbered as '1', '2', '3', and '4' the four possible collinear magnetic structures with $\mathbf{k} = (0,0,0)$, as described in Table 8. The observed magnetic structure is $(F_X^{(1)}, -F_X^{(2)}) = (++;$ - -) and is labeled as '4' in the diagram of Figure 12. The incommensurate structure and the domains where no classical order can be found by the program are labeled as '5'. Then, we get the conditions to obtain the observed magnetic state $[\mathbf{k} = (0,0,0)$ and the spin sequence $(F_X^{(1)}, -F_X^{(2)}) = (++; --)]$ as the magnetic ground state. The phase diagram is much simpler than that for A-LFP (Figure 12). The different domains can be distinguished only by the sign of the exchange integrals, not by their strength. If $J_1 > 0$, the observed magnetic structure is not found as ground state. It is only observed when $J_1 < 0$ and $J_2 > 0$ and $J_3 > 0$. Note that a positive value of J_2 can be understood from the Goodenough–Kanamori–Anderson prediction rules^{36,37} extended to the super-super exchange interactions, because this exchange path exhibits the smallest angles, and also should be less antiferromagnetic than the others. The conditions to observe the other collinear structures are reported in Table 8.

Remarks and Conclusion

We have studied the crystal and magnetic structures of two iron phosphates: $A-Li_3Fe_2(PO_4)_3$ and $B-Li_3Fe_2$ -(PO₄)₃ (NASICON). Both of them show long-range magnetic ordering at low temperature. In both structures, the magnetic moments of iron atoms are antiparallel in a same lantern unit. We have determined the conditions to be satisfied by super-super exchange integrals to find the observed magnetic structure as the magnetic ground state. It is expected that for all compounds involving lantern units the exchange is negative and will lead to a global antiferromagnetic behavior.

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⁽³⁶⁾ Goodenough. J. B. *Magnetism and the Chemical Bond*; Interscience Publishers: New York, 1963.

⁽³⁷⁾ Kanamori, J. J. Phys. Chem. Solids 1959, 10, 87.